



7. Of the following pairs which would be the larger of the two.
- Na/Na<sup>+</sup>
  - F/F<sup>-</sup>
  - Al/Al<sup>+3</sup>
  - N/N<sup>-3</sup>
8. Which element would have the lowest electronegativity?
- Cl, F
  - C, N
  - Mg, Ne
  - As, Ca
9. Label the following attached periodic table with the following terms:
- Row/Period
  - Group/Family/Column
  - Metal
  - Nonmetal
  - Metalloid
  - Alkaline metals
  - Alkaline Earth metals
  - Halogens
  - Noble Gases
  - Transition metals
  - Inner transition metals
  - s-block
  - p-block
  - d-block
  - f-block
  - lanthanide series
  - actinide series