

Polyatomic Compounds

From: <http://dbhs.wvusd.k12.ca.us/webdocs/Nomenclature/Polyatomic-FormulatoName.html>

And <http://dbhs.wvusd.k12.ca.us/webdocs/Nomenclature/Polyatomic-NametoFormula.html>

Write the correct name for:

- 1) KClO_3
- 2) SnSO_4
- 3) $\text{Al}(\text{MnO}_4)_3$
- 4) $\text{Pb}(\text{NO}_3)_2$
- 5) $\text{Mg}_3(\text{PO}_4)_2$
- 6) $\text{Fe}(\text{HCO}_3)_3$
- 7) Na_2CO_3
- 8) MnSO_4

Write the correct formula for:

- 9) potassium perchlorate
- 10) lead (IV) phosphate
- 11) aluminum sulfate
- 12) iron(II) carbonate
- 13) barium iodate
- 14) tin(II) sulfide
- 15) plumbous cyanide
- 16) silver phosphate
- 17) cobalt(III) nitrite

and three special additions:

- 18) ammonium sulfate
- 19) hydrochloric acid
- 20) chloric acid